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APPLICATION NO.	FILING DATE	FIRST NAMED INVENTOR	ATTORNEY DOCKET NO.	CONFIRMATION NO.
10/598,958	09/15/2006	Hidetoshi Ito	112857-607	2591
	7590 01/13/200 & LLOYD, LLP	EXAMINER		
P. O. BOX 113:	5	HOBBS, LISA JOE		
CHICAGO, IL 60690			ART UNIT	PAPER NUMBER
			1657	
			MAIL DATE	DELIVERY MODE
			01/13/2009	PAPER

Please find below and/or attached an Office communication concerning this application or proceeding.

The time period for reply, if any, is set in the attached communication.

Advisory Action Before the Filing of an Appeal Brief

Application No.	Applicant(s)	
10/598,958	ITO ET AL.	
Examiner	Art Unit	

	Lisa J. Hobbs	1657					
The MAILING DATE of this communication appe	ars on the cover sheet with the c	correspondence add	ress				
THE REPLY FILED 19 December 2008 FAILS TO PLACE THIS APPLICATION IN CONDITION FOR ALLOWANCE.							
1. The reply was filed after a final rejection, but prior to or on application, applicant must timely file one of the following application in condition for allowance; (2) a Notice of Appetor Continued Examination (RCE) in compliance with 37 C periods:	replies: (1) an amendment, affidavit eal (with appeal fee) in compliance	t, or other evidence, w with 37 CFR 41.31; or	hich places the (3) a Request				
a) The period for reply expires 3 months from the mailing date b) The period for reply expires on: (1) the mailing date of this A no event, however, will the statutory period for reply expire to Examiner Note: If box 1 is checked, check either box (a) or (MONTHS OF THE FINAL REJECTION. See MPEP 706.07(f)	dvisory Action, or (2) the date set forth in ater than SIX MONTHS from the mailing b). ONLY CHECK BOX (b) WHEN THE	date of the final rejection	n.				
Extensions of time may be obtained under 37 CFR 1.136(a). The date of have been filed is the date for purposes of determining the period of extunder 37 CFR 1.17(a) is calculated from: (1) the expiration date of the set forth in (b) above, if checked. Any reply received by the Office later may reduce any earned patent term adjustment. See 37 CFR 1.704(b). NOTICE OF APPEAL	ension and the corresponding amount of hortened statutory period for reply original for replacements or reply original for replacements or re	of the fee. The appropria nally set in the final Office	ate extension fee e action; or (2) as				
2. The Notice of Appeal was filed on A brief in comp filing the Notice of Appeal (37 CFR 41.37(a)), or any exter Notice of Appeal has been filed, any reply must be filed wi	nsion thereof (37 CFR 41.37(e)), to	avoid dismissal of the					
3. The proposed amendment(s) filed after a final rejection, be a considered and amendment(s) filed after a final rejection, be a considered amendment(s) filed after a final rejection, be a considered and a considered amendment and a cons	nsideration and/or search (see NOTw); ter form for appeal by materially rec	E below); ducing or simplifying the					
NOTE: (See 37 CFR 1.116 and 41.33(a)). 4. The amendments are not in compliance with 37 CFR 1.12 5. Applicant's reply has overcome the following rejection(s): 6. Newly proposed or amended claim(s) would be all							
non-allowable claim(s). 7. For purposes of appeal, the proposed amendment(s): a) how the new or amended claims would be rejected is proved the status of the claim(s) is (or will be) as follows: Claim(s) allowed: Claim(s) objected to: Claim(s) rejected: Claim(s) withdrawn from consideration: AFFIDAVIT OR OTHER EVIDENCE		be entered and an ex	cplanation of				
 The affidavit or other evidence filed after a final action, but because applicant failed to provide a showing of good and was not earlier presented. See 37 CFR 1.116(e). 							
9. The affidavit or other evidence filed after the date of filing entered because the affidavit or other evidence failed to o showing a good and sufficient reasons why it is necessary 10. The affidavit or other evidence is entered. An evidence is entered.	vercome <u>all</u> rejections under appea and was not earlier presented. Se	ll and/or appellant fails ee 37 CFR 41.33(d)(1)	s to provide a				
 The affidavit or other evidence is entered. An explanation <u>REQUEST FOR RECONSIDERATION/OTHER</u> The request for reconsideration has been considered but 		•					
See Continuation Sheet.		CONTRIBUTION ANDWAIN	oc pecause.				
12. ☐ Note the attached Information <i>Disclosure Statement</i>(s). (13. ☐ Other:	r 1 0/56/06) raper No(s)						
	/Lisa J. Hobbs/ Primary Examiner Art Unit: 1657						

U.S. Patent and Trademark Office PTOL-303 (Rev. 08-06) Continuation of 11. does NOT place the application in condition for allowance because:

First, the rejection of claims 11-20 on the ground of nonstatutory double patenting over claims 14-26 of copending Application No. 10/536,934 is not addressed in the response of 19 December 2008.

Second, applicant's arguments regarding the rejection of claims 11-20 under 35 USC 103(a) has been fully considered and is not deemed persuasive. Applicant argues that the prior art references teach the hydrolysis of water, not the system as disclosed by applicant using ionized cross-linked polyacrylamide gels, which avoid the release of gases. However, Shahinpoor teaches at col. 1, lines 26-28, that that "the creation of sensors and controllable actuators, or synthetic muscles, is known. Sensors and artificial muscles or actuators made from ion-exchange membranes are relatively new but known" and that the polymeric hydrogel components taught by applicants, see Example 1, are known "U.S. Pat. No. 5.100.933, to Tanaka, et al., discloses the use of ionized cross-linked polyacrylamide gels as engines or artificial muscles; the gels can contain a metal ion and are capable of discontinuous volume changes induced by infinitesimal changes in environment. The gel is made by dissolving acrylamide monomers and bisacrylamide monomers in water, adding a polymerization initiator (in particular, ammonium persulfate and TEMED, or tetramethyl-ethylene-diamine) to the solution, soaking the gel sample in water to wash away all residual monomers and initiators, immersing the gel in a basic solution of TEMED for up to 60 days, then immersing the gel in a solvent (in particular, acetone, acetone in water, ethanol and water, or methanol and water). The primary disadvantages of these actuators are generally that the response time of the gel is much longer than that of other known actuator components and that the gel must be contained in the solvent bath. The gels are also mechanically brittle and easily broken" (col. 2, lines 3-20). Also, Madden et al. teach that the "electrolyte may be a liquid (which may require actuator encapsulation), a gel, or a solid"...specifically, the electrolyte may be a polymethylmethacrylate (PMMA) gel containing a salt dopant (col. 4, lines 15-20. As well, Hirai et al. teach that in addition to the PVA gel swollen with DMSO, they were able to achieve gel deformation and motive force using PVC and poly(methyl methacrylate) (p. 199).

Applicants argue that Adolf does not teach the palladium catalyst or the coil/mesh structure, however, Madden et al. teach electrodes of gold, palladium, platinum, etc., (col. 6 and claims 21-23) and Shahinpoor teaches electrodes of noble metals, including platinum, palladium, and nickel (col. 6) and Adolf et al. teach that platinum may be used in the actuator (col. 7). Madden et al. specifically teach that the electrode comprises a coiled metal wire (claim 16).

Applicants argue that each of the references teaches away from the idea of the instant invention, however, each of the references, while disclosing multiple and various types and species of polymer actuators, discloses that polymer gel actuators are well known in the art, several teach that the use of cross-linked gel actuators comprising acrylaminde components are known, several teach that electrodes of metal, such as palladium, are known, and several teach that electrodes in various configurations such as mesh and coil are known. Finally, each teaches that one of skill in the art knows how to place the electrodes to achieve optimal activity for the desired reaction of that, particular gel actuator.